October 2021

UNDERGRADUATE RESEARCH, CHEMISTRY 1710

Instructions: Select areas that have the greatest appeal to you; then make appointments with the research directors to discuss the specific projects they have available. [Undergraduate research is an important learning experience that should not be too time dependent. However, since you are taking other classes, be sure to ask each research director approximately how much laboratory time per credit he/she will expect of you.] After deciding upon the project you wish to undertake, check with the faculty member concerned to make sure he/she agrees to accept you.

Since a permission number is necessary for registering for Chemistry 1710 you must stop in Room 107 Chevron Science Center and see **George C. Bandik** or **Sharon Mansfield** and also ***let us know whom you have selected as a research director; be sure to give the name of this individual. We need this information so that a grade report can be issued at the end of the term.***

Director	Research Area
S. AMEMIYA (803A CHVRN-41217) amemiya@pitt.edu	Bioanalytical Chemistry and Electrochemistry: Development, miniaturization and theory of electrochemical sensors, and application of these as a probe of scanning electrochemical microscope in order to understand molecular transport at biological membranes. **http://www.chem.pitt.edu/person/shigeru-amemiya**
S. A. ASHER (701 CHVRN-48570) asher@pitt.edu	Development of new laser spectroscopic techniques and optical devices. Resonance Raman and vibrational studies of peptide and protein folding. Development of new smart materials for chemical sensing, nonlinear optics and display devices. **http://www.chem.pitt.edu/person/sandy-asher**
K. M. BRUMMOND (807 CHVRN-41955) kbrummon@pitt.edu	Research in our laboratory focuses on the discovery and development of synthetic transformations to facilitate and overcome the challenges posed by the preparation of molecularly complex bioactive compounds. **http://www.chem.pitt.edu/person/kay-m-brummond**
L. T. CHONG (331 EBERL- 46026) ltchong@pitt.edu	Computational Biophysical Chemistry. We apply computer simulations to study the mechanisms of complex biological processes such as protein binding and switching at the atomistic level. Areas of interest include: (i) development of weighted ensemble methods and software for extending the timescales of the computer simulations, (ii) development of more accurate simulation models, (iii) rational design of protein-based sensors for detecting ligands of interest, and (iv) estimating rate constants for receptor-ligand unbinding processes. Especially interested in undergraduates who would like to do research for at least a year, including full-time research during the summer. **http://www.chem.pitt.edu/person/lillian-chong**

R. COALSON
(321 EBERL-48261)
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Theory of chemical dynamics, with applications to optical spectroscopy (absorption, Raman scattering, photo-dissociation processes), tunneling, isomerization, optical fiber design, neutron scattering cross sections, and other experimentally observable phenomena. Emphasis is on using measured spectra and cross sections to unravel microscopic mechanisms.
http://www.chem.pitt.edu/person/rob-coalson

D. P. CURRAN
(1101 CHVRN-48240)
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New strategies for the total synthesis of structurally interesting and/or biologically active natural products and the development of new synthetic methodology. Radical chemistry. Fluorous chemistry.
http://www.chem.pitt.edu/person/dennis-curran

A. DEITERS
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Chemical Biology: We are developing new chemical tools to study biological processes relevant to human health. Our multidisciplinary research program involves organic synthesis, cell and molecular biology, protein engineering, nucleotide chemistry, amino acid chemistry, photochemistry, as well as medicinal and organometallic chemistry.
http://www.pitt.edu/~deiters/

P. E. FLOREANCIG
(1203 CHVRN-48727)
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Organic chemistry: Development of methodology for the construction of carbon-carbon bonds and unusual functionality. Synthesis of small molecules designed to elucidate mechanistic details of biochemical phenomena. Natural products synthesis.
http://www.chem.pitt.edu/person/paul-floreancig

S. GARRETT-ROE
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CO_2 ABSORPTION BY IONIC LIQUIDS: Global warming from fossil fuel use is a major threat to the environment. Currently technologies to capture CO_2 from the exhaust of fossil-fuel burning power plants are economically infeasible. Ionic liquids might change that. Ionic liquids are fluids made exclusively of cations and anions, and whose properties can be tuned for particular tasks. Ionic liquids are being developed to capture CO_2 more efficiently, but progress is limited by our poor understanding of the molecular interactions between the CO_2 and the ionic liquid. To address the fundamental interactions between CO_2 and the ionic liquids, we use vibrational spectroscopy of the antisymmetric stretch of CO_2. To explore these issues, students will load CO_2 into a series of ionic liquids and measure FTIR spectra. The students will diagnose cooperative interactions with the cation and anion by determining if the vibrational frequency shifts of the CO_2 across a series of ionic liquids that vary the cation and anion are or are not additive. The project will advance our understanding of how this potentially useful class of solvents absorb CO_2.
http://www.chem.pitt.edu/person/sean-garrett-roe

R. HERNÁNDEZ SÁNCHEZ
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We are a research group focused in combining supramolecular, inorganic, and materials chemistry to synthesize functional systems that bridge the gap between nanoscale materials and molecular chemistry. **https://www.chem.pitt.edu/person/raul-hernandez-sanchez**

W. SETH HORNE
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Research in the Horne lab is focused on the design, synthesis, and study of synthetic analogues of proteins. These non-natural species can provide insight into the behavior of proteins found in nature and can also act as scaffolds for the design of molecules with interesting structures and functions. Our research takes place at the interface between organic chemistry, biochemistry, biophysics, structural biology and materials science.
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G. HUTCHISON
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Our group develops new materials, as well as microscale and nanoscale functional devices literally from the bottom up. We focus on building electronic materials from molecular subunits, both organic and inorganic, using a variety of techniques to rationally design the desired properties. This encompasses chemical synthesis, characterization (both physical and chemical), combined with theoretical modeling and simulation. Areas of research cover: Materials and Nanoscale Chemistry. Computational Materials Design. Rational Design and Materials Synthesis. Electronic Materials. Nanoscale Dynamics.
http://www.chem.pitt.edu/person/geoffrey-hutchison

K. D. JORDAN
(330 EBERL-48690)
jordan@pitt.edu

Theoretical studies of molecular structure and reaction mechanisms. Computer simulations with emphasis on hydrogen-bonded systems, chemical reactions on surfaces, and characterization of biomolecules. We are also applying computer simulations to address a range of problems in the energy area.
http://www.chem.pitt.edu/person/kenneth-jordan

K. KOIDE
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We design, synthesize, and apply synthetic organic molecules for various problems. For example, we chemically synthesize complex natural products and study them in cancer. In addition, we develop fluorescent chemosensors for biological imaging in cells and animals and for the quality control of drinking water, environment, and discoveries of precious metals.
http://www.chem.pitt.edu/person/kazunori-koide

J. LAASER
(G12A CHVRN-30125)
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The Laaser lab works at the boundary between polymer chemistry and polymer physics, exploiting both controlled synthesis methods and physical and optical characterization techniques to understand how molecular-level interactions between polymer chains dictate the bulk properties of these materials. We are particularly interested in polymer systems containing strong ionic interactions between chains, and in polymer systems that respond to external stimuli like mechanical force. Depending on current projects, opportunities may be available for undergraduate researchers interested in polymer synthesis, polymer characterization, spectroscopy, and/or computational modelling.

http://laaserlab.chem.pitt.edu

H. LIU
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hliu@pitt.edu

Research in the H. Liu lab is focused on the chemistry and applications DNA and carbon materials. DNA Nanotechnology: we use DNA to prepare well-define nanoscale objects and explore their use in nanoscale patterning. The targeted applications are the fabrication of nanoelectronics and antifouling/self-cleaning surfaces. Carbon Materials: we study the intrinsic surface properties of carbon materials. Our recent work showed that many surface properties of carbon materials are masked by surface contamination by airborne volatile organic compounds. Removing these contaminations resulted drastic improvement in material performances. We are exploring the effect of surface contamination on applications related to environmental science, energy storage, and composite materials.
http://www.chem.pitt.edu/person/haitao-liu

P. LIU
(225 EBERL-35065)
pengliu@pitt.edu

Computational Organic Chemistry: The Liu group use computational tools to study organic and organometallic reactions. We use quantum mechanical calculations to study how reactions occur, factors controlling rates and selectivity, and provide theoretical insights to help develop improved catalysts and reagents. We are also developing a multi-scale computational screening protocol to facilitate the discovery of new catalysts.
http://www.chem.pitt.edu/people/faculty/peng-liu

T. Y. MEYER
(1003 CHVRN-48635)
tmeyer@pitt.edu

Polymers and Materials. The synthesis and characterization of Repeating Sequence Copolymers (RSCs) for electronic and bio-medical applications. The design and preparation of stimuli-responsive hydrogels. Especially interested in undergraduates who would like to do research for two semesters or more.
http://www.chem.pitt.edu/person/tara-meyer

J. MILLSTONE
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Inorganic and Materials Chemistry; Nanomaterials; Surface and Colloid Chemistry. Whether they will be used in catalysis or artificial limbs, nanoparticle surfaces influence every aspect of their behavior. For example, the ligand shell of a nanocrystal can determine its luminescence, its performance in a solar cell, or its clearance from the human body. In the Millstone group, we are interested in synthetically controlling both the crystallographic and chemical composition of the nanoparticle surface, in order to develop new nanoparticle architectures that will have applications in fields ranging from catalysis to medicine.
http://www.chem.pitt.edu/person/jill-millstone

S. G. NELSON
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Natural Products Total Synthesis and Asymmetric Catalysis. Development of new synthetic methodology for the rapid and efficient construction of therapeutically relevant natural products. Design and execution of new asymmetric catalytic procedures for stereoselective carbon-carbon bond constructions.
http://www.chem.pitt.edu/person/scott-nelson

N. L. ROSI
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nrosi@pitt.edu

Nanoscience and Materials Chemistry. We employ a variety of techniques from the traditional disciplines of modern chemistry to develop new methods of assembling and organizing molecular and nanoparticle building blocks into functional materials, such as catalysts, sensors, and gas storage containers.
http://www.chem.pitt.edu/person/nathaniel-rosi

S. K. SAXENA
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Biophysical Chemistry: Conformational dynamics in proteins, and protein interactions and structures are studied using Fourier Transform electron spin resonance. New spectroscopic methods and instrumentation are also developed.
http://www.chem.pitt.edu/person/sunil-saxena

A. STAR
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Star group research interests are in areas of molecular recognition at nanoscale and nanotechnology enabled chemical and biological sensing. This research, in addition to basic understanding of chemical and biological processes, will yield a prospect of building novel detection methods for widest possible range of chemical and biological analytes. Our group is currently working with a commercial partner to develop a disposable nanoelectronic sensor for the measurement of carbon dioxide (CO_2) and nitric oxide (NO) in human respiration. These disposable nanoelectronic sensors will extend the reach of the gas monitoring into emergency medicine and point-of-care settings.
http://www.chem.pitt.edu/person/alexander-star

D. H. WALDECK
(G-10 CHVRN-48430)
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Experimental Physical Chemistry, with applications to electron transport and electron transfer in nanometer scale assemblies. Our group is using electrochemistry, photochemistry, and electrical measurements to examine how the chirality of molecules and materials affect the charge and spin current in electron transfer processes. This work has implications for spintronics, enantioselective chemistry, and homochirality in biology.
http://www.chem.pitt.edu/person/david-waldeck

Y. WANG
(505 CHVRN-40028)
ym.wang@pitt.edu

Organic synthesis, organometallic chemistry and catalysis. The Wang group works on the development of new catalysts, reactions, and reagents for the synthesis of high-value molecular building block from readily accessed starting materials.
https://www.chem.pitt.edu/person/yiming-wang

M. WARD
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Undergraduates can participate in research related to the development of new curriculum for upper-level analytical laboratory courses.
www.chem.pitt.edu/person/michelle-m-ward

S. G. WEBER
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We address the challenges of acquiring chemical information of living systems through devising and applying measurement concepts with unique attributes. Much of our work involves a need to understand or control mass transport by diffusion, convection, and electroosmosis/electrophoresis. Recently, we have created systems for making quantitative measurements of neurotransmitters *in vivo* by microdialysis/fast capillary LC, for electroosmotically perfusing tissue to determine neuropeptide hydrolysis rates. Core analytical techniques are electrochemistry and liquid chromatography.
http://www.chem.pitt.edu/person/steve-weber

P. WIPF
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Organic chemistry: Total synthesis and study of biologically important molecules and novel, reactive functionalities. New technologies, including the use of 3D printing of synthesis cassettes and blue LED powered photochemistry in-flow.
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